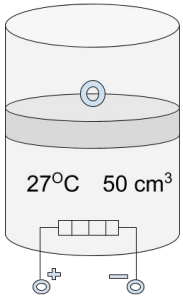


Using the Gas Equation

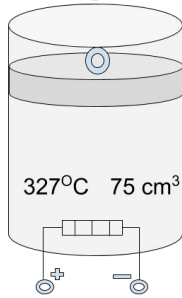
$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Use the gas equation to find either the new pressure, volume or temperature

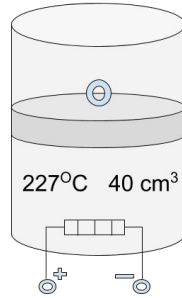
1 250 k Pa



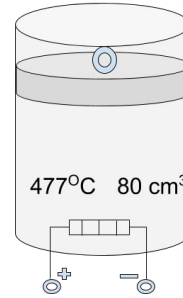
P_2



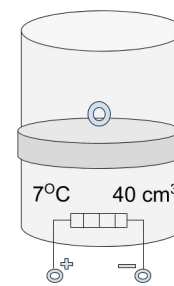
2 250 k Pa



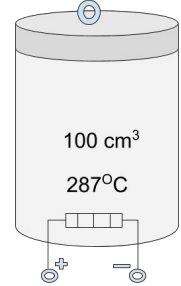
P_2



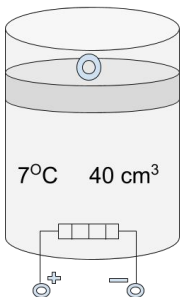
3 100 k Pa



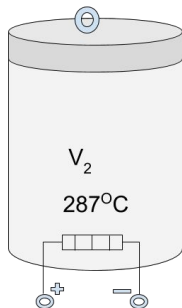
P_2



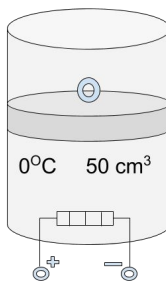
4 100 k Pa



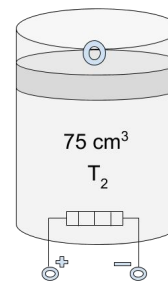
80 kPa



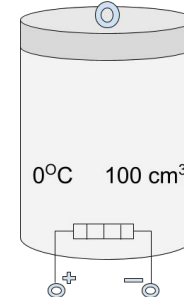
5 100 k Pa



200 kPa



6 100 kPa



400 kPa

